Blends of Pheromones, With and Without Host Plant Volatiles, Can Attract Multiple Species of Cerambycid Beetles Simultaneously

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Abstract

Pheromone components of cerambycid beetles are often conserved, with a given compound serving as a pheromone component for multiple related species, including species native to different continents. Consequently, a single synthesized compound may attract multiple species to a trap simultaneously. Furthermore, our previous research in east-central Illinois had demonstrated that pheromones of different species can be combined to attract an even greater diversity of species. Here, we describe the results of field bioassays in the northeastern, midwestern, south-central, and southwestern United States that assessed attraction of cerambycids to a 'generic' pheromone blend containing six known cerambycid pheromone components, versus the individual components of the blend, and how attraction was influenced by plant volatiles. Nineteen species were attracted in significant numbers, with the pheromone blend attracting about twice as many species as any of the individual components. The blend attracted species of three subfamilies, whereas individual components were identified. The plant volatiles ethanol and α -pinene usually enhanced attraction to the blend. Taken together, these experiments suggest that blends of cerambycid pheromones, if selected carefully to minimize inhibitory effects, can be effective for sampling a diversity of species, and that plant volatiles generally enhance attraction. Such generic pheromone blends may serve as an effective and economical method of detecting incursions of exotic, potentially invasive species.

Key words: chemical ecology, pheromones, monitoring

The large beetle family Cerambycidae includes many species that are among the most important insect pests of woody plants in natural and managed ecosystems worldwide (Haack 2017, Wang 2017). Because the larvae of these species develop within wood and wooden products, they are readily transported around the world by international commerce and so are among the most common potentially invasive species intercepted in international quarantine (Eyre and Haack 2017). Traps baited with synthesized pheromones offer a means of monitoring for new incursions of exotic species of cerambycids, as well as for managing populations of native pest species, and monitoring threatened species (e.g., Rassati et al. 2012, Dutcher and Bactawar 2016, Larsson 2017, Sweeney et al. 2017). Research to date has suggested that cerambycids use either maleor female-produced attractant pheromones whose function appears to break out along subfamily lines. Thus, aggregation-sex pheromones, produced by males and attracting both sexes, are used by species in the subfamilies Cerambycinae, Lamiinae, and Spondylidinae, whereas female-produced sex pheromones that attract only males are found in the Prioninae and Lepturinae (Hanks and Millar 2016). There also is growing evidence that several pheromone structures have been conserved within the family and are shared among many related species that may be sympatric, or native to different continents (Millar and Hanks 2017). For example, the aggregation-sex pheromones of many cerambycine species comprise individual or blends of stereoisomers of

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3-hydroxyalkan-2-ones and/or the related 2,3-alkanediols, whereas those of lamiines and spondylidines are often composed of enantiomers of fuscumol ([*E*]-6,10-dimethylundeca-5,9-dien-2-ol) and/or fuscumol acetate ([*E*]-6,10-dimethylundeca-5,9-dien-2-yl acetate), or monochamol (2-[undecyloxy]ethanol; Hanks and Millar 2016). Similarly, the female-produced sex pheromone prionic acid ([*3R*,*5S*]-3,*S*-dimethyldodecanoic acid) appears to be shared by at least several genera of prionines in different parts of the world (Barbour et al. 2011, Wickham et al. 2016).

Because of the conservation of pheromone structures within the Cerambycidae, it often is possible to attract multiple species to singlecomponent lures (Hanks et al. 2012, Sweeney et al. 2014, Wickham et al. 2014, Miller et al. 2017). The number and diversity of species that are attracted may be increased further by combining the pheromones of multiple species. Such blends could be advantageous for surveillance monitoring for exotic species, because they would reduce the number of traps necessary to attract species of multiple subfamilies and so reduce costs for materials and labor. In addition, blends may be advantageous because some cerambycid species respond most strongly to synergistic combinations of their pheromone components, and less so or not at all to the individual components (e.g., Lacey et al. 2009, Meier et al. 2016, Zou et al. 2016, Miller et al. 2017). Conversely, chemicals which are not components of the pheromone of a particular species may antagonize attraction of that species. Such antagonism may be an adaptation to avert unproductive attraction to pheromones of heterospecifics (Mitchell et al. 2015, Meier et al. 2016).

As a first step toward devising a multipurpose mixture of synthesized pheromones which would attract the maximum number of cerambycid species, we previously had conducted field tests in east-central Illinois with a blend of three chemicals that are typical pheromones of cerambycines (racemic 3-hydroxyhexan-2-one, syn-2,3-hexanediol, racemic 2-methylbutan-1-ol) with three common pheromone components of lamiines (fuscumol, fuscumol acetate, monochamol), versus the individual components (Hanks et al. 2012). Those experiments demonstrated that attraction of a given species to its dominant pheromone components in the blend may indeed be affected by the presence of other components, with attraction of several species being either synergized or antagonized by certain components. A second experiment revealed that attraction to the pheromone blend usually was enhanced by the host plant volatiles ethanol and α-pinene. In an independent study, the two experiments were replicated in counties across the state of Pennsylvania (Hanks and Millar 2013), with more than 30 cerambycid species responding to particular treatments.

Here, we describe results from conducting the same two experiments in a number of widely separated geographic regions of the United States, with the goal of more broadly assessing the efficacy of the same pheromone blend in attracting multiple cerambycid species, with and without host plant volatiles as potential synergists. The experiments were replicated independently at study sites in the northeastern (New York), midwestern (Michigan, Indiana), southeastern (Florida), south-central (Texas), and southwestern (California) United States. A companion article (Millar et al. 2017) summarizes complementary field bioassays in which some of the same cerambycid pheromones were field tested individually and in binary blends at some of the same study sites.

Materials and Methods

Sources of Chemicals

Compounds purchased from commercial sources included racemic 3-hydroxyhexan-2-one (henceforth '3-ketol'), racemic fuscumol, racemic fuscumol acetate, and monochamol (all from Bedoukian Research, Inc., Danbury, CT), and racemic 2-methylbutan-1-ol (Aldrich Chemical, Milwaukee, WI). Racemic *syn*-2,3-hexanediol ('*syn*-diol') was synthesized as described in Lacey et al. (2004), and prionic acid as described in Rodstein et al. (2009). See General Methods of Trapping section below for doses and blends of these compounds. Plant volatiles were emitted from high-release lures (95% ethanol, ~0.4 g/d, model IP036-100; α -pinene, ~2 g/d, model IP037-75; Synergy Semiochemical Corp., Burnaby, British Columbia, Canada).

Study Sites

Suitable sites for bioassays were located by collaborators in the six states, consisting of stands of mixed hardwood and/or coniferous trees in natural or managed environments (Table 1). The number of transects per experiment and duration of experiments were subject to the logistical and time constraints of the individual collaborators (Table 1).

General Methods of Trapping

Bioassays at all study sites used cross-vane panel traps (black corrugated plastic; AlphaScents, Portland, OR) that had been coated with a liquid fluoropolymer lubricant (Fluon, Northern Products, Inc., Woonsocket, RI) to improve trapping efficiency (Graham et al. 2010). In most cases, trap basins contained either diluted propylene glycol or saturated aqueous NaCl with a few drops of detergent to kill and preserve captured beetles. However, at the Florida site the basins contained only water with detergent, because traps usually were checked for beetles every few days.

Trap lures were polyethylene sachets (press-seal bags, Bagette model 14770, 5.1 × 7.6 cm, 0.05-mm thick, Cousin Corp., Largo, FL) that were loaded individually with the six test compounds, or the same six compounds combined in a blend, in both cases using 50 mg of the racemic compounds (i.e., 25 mg of each enantiomer of 3-ketol, syn-diol, 2-methylbutan-1-ol, fuscumol, and fuscumol acetate) or 25 mg of the achiral monochamol, in 1 ml of isopropanol (estimated release rates: ~0.77, 0.4, 0.22, 0.4, 0.48, and 0.07 mg/d, respectively). However, individual lures for the sex pheromone prionic acid were loaded with only 1 µl of the synthesized chemical in 1 ml of isopropanol (release rate ~1.8 µg/d), because even minute doses of that compound are highly attractive to Prionus species (Rodstein et al. 2011). Solvent control lures contained 1 ml of isopropanol. At most of the sites, traps were suspended <2 m above the ground from steel poles or frames of polyvinyl chloride irrigation pipe. The exceptions were two sites where traps were suspended from tree branches: Texas (traps <2 m above the ground) and California (traps at least 4 m above the ground to prevent damage by bears).

Traps were deployed 5–15 m apart in linear transects with treatments assigned randomly to traps on the day of setup, with one trap per treatment in each transect (see Table 1 for dates of experiments). Treatments for experiment 1 were as follows: 1) 3-ketol, 2) *syn*-diol, 3) 2-methylbutan-1-ol, 4) fuscumol, 5) fuscumol acetate, 6) monochamol, 7) the complete blend of these compounds, and 8) solvent control. At the study site in Florida, fuscumol and fuscumol acetate were blended together, and not tested separately, due to the limited space available for the transect.

Treatments for experiment 2 were as follows: 1) the same blend of six pheromone components, 2) the pheromone blend with the plant volatiles ethanol + α -pinene, 3) plant volatiles alone, and 4) solvent control, with one trap per treatment in each transect. Both experiments were conducted at study sites in New York, Michigan, and Florida, experiment 1 only in Indiana and Texas, and experiment 2 only in California (Table 1). The number of trap transects

| Table 1. Study sites for two field experiments that tested attraction of cerambycid beetles to a blend of synthesized pheromones (experi- |
|---|
| ment 1) and the influence of plant volatiles (experiment 2) in various areas of the United States during 2011, and the nature of the sur- |
| rounding forests |

| State, experiment, study site | GPS (lat., long.) | Habitat type |
|--|-------------------|---|
| New York | | |
| Experiment 1 (25 May to 20 Aug.) and 2 (16 June to 20 Aug.) | | |
| SUNY-ESF James F. Dubuar Memorial For., Adirondack Park, | 44.163, -74.908 | Mature hardwoods, managed conifers |
| St. Lawrence Co. | | |
| Frank E. Jadwin State For., Lewis Co. | 44.076, -75.381 | Mature northern hardwoods |
| SUNY-ESF Lafayette Road Field Station, Onondaga Co. | 42.991, -76.132 | Arboretum: many hardwood and conifer species |
| SUNY-ESF Heiberg Memorial For., Cortland Co. | 42.768, -76.072 | Mixed hardwoods and managed conifers |
| Allegany State Park, Cattaraugus Co. | 42.091, -78.851 | Mixed hardwoods and conifers |
| Michigan | | |
| Experiment 1 (23 Jun. to 9 Aug.) | | |
| Michigan State Univ. Tree Res. Center, Lansing, Ingham Co. | 42.672, -84.475 | Mixed hardwoods |
| Experiment 2 (21 Jun. to 11 Aug.) | | |
| MSU W. K. Kellogg Experimental Forest, Kalamazoo Co. | 42.363, -85.358 | Mixed hardwoods and conifers |
| Lakeshore Park, Novi, Oakland Co. | 42.511, -83.485 | Mixed hardwoods and conifers |
| Tollgate Educational Center, Novi, Oakland Co. | 42.499, -83.459 | Mixed hardwoods |
| Indiana | | |
| Experiment 1 (10 May to 9 Aug.) | | |
| Martell For., Tippecanoe Co. (2 sites) | 40.435, -87.034; | Mixed hardwoods |
| | 40.442, -87.035 | |
| Florida | | |
| Experiments 1 (24 Apr. to 31 Jul.) and 2 (7 Aug. to 16 Oct.) | | |
| Private residence, Port St. Lucie, St. Lucie Co. | 27.231, -80.383 | Urban landscape with hardwoods and conifers of many |
| | | species |
| Texas | | |
| Experiment 1 (25 Apr. to 5 Jul.) | | |
| Brackenridge Field Lab., Travis Co. | 30.281, -97.777 | Mixed hardwoods |
| Stengl Field Lab., Bastrop Co. | 30.089, -97.166 | Mixed hardwoods and conifers |
| California | | |
| Experiment 2 (20 Aug. to 16 Oct.) | | |
| Slaughterhouse Ravine, Butte Co. | 39.839, -121.617 | Mixed oak-conifer |
| Childs Meadows, Tehama Co. | 40.345, -121.484 | Conifers |

Study sites are ordered so as to progress from states in the northeast to the midwest, south, and west.

per experiment varied from one to four, and transects within a site were widely separated (Table 1). Insects were collected from traps at intervals of 3–14 d, and on those days, treatments were rotated down transects to control for positional effects.

In addition to experiments 1 and 2, a single sentinel trap baited with prionic acid was deployed at study sites in New York, Michigan, Indiana, Florida, and Texas to determine whether there were *Prionus* species present (dates as in Table 1). This prionic acid treatment was not included in the statistical analysis of treatment effects.

Statistical Analysis

For individual species at each study site, overall treatment effects were tested with the nonparametric Friedman's test (PROC FREQ, option CMH; SAS Institute 2011), with the number of replicates based on the number of trap transects (if more than one) and the number of collection dates. Replicates that contained no specimens of the species in question in any of the treatments, which would contribute nothing to testing treatment effects, were dropped from analyses. Data were analyzed only for species that were represented by at least eight specimens at a study site, the minimum number that could result in statistically significant treatment effects. Although the collection date term was highly significant in each analysis, it is not reported here, because trap catches invariably changed with date due to the vagaries of weather and other abiotic and biotic factors. Pairwise comparisons of treatments were tested with the REGWQ test

(controlling experiment-wise error rates; SAS Institute 2011) and were protected (i.e., assuming a significant overall Friedman's test).

Captured beetles were identified according to Monné and Hovore (2005) and Lingafelter (2007). Voucher specimens of the captured species were retained by the individual collaborators (those from New York and Florida are available from the laboratory collection of LMH).

Results and Discussion

Bioassays in the six different states resulted in capture of 6,539 cerambycid beetles and a few beetles from the related family Disteniidae (Supp. Table 1 [online only]). These comprised 142 species of cerambycids, including 48 species in 14 tribes of the subfamily Cerambycinae, 55 species in 10 tribes of the Lamiinae, 27 species in 2 tribes of the Lepturinae, 4 species in 2 tribes of the Prioninae, 7 species in 1 tribe of the Spondylidinae, and 1 species of the Parandrinae. Traps in Indiana and Texas also caught 11 adults of the disteniid Elytrimitatrix undata (F.). The greater species diversity and/ or numbers per species of cerambycines, lamiines, and spondylidines is due to the fact that bioassay treatments included common pheromone components of species in those subfamilies but no known pheromones of lepturines, parandrines, or disteniids (see below in this section). Sentinel traps baited with prionic acid caught males of Prionus imbricornis (L.) in Texas and Prionus pocularis Dalman in Florida, but no prionines were caught in New York, Michigan, or

Indiana, possibly due to the timing of bioassays in relation to flight periods (Supp. Table 1 [online only]).

For Experiment 1, there were 32 cases of significant treatment effects, with some of the more common species being attracted to the same compounds in multiple states (Table 2). Pheromones, or likely pheromones (i.e., confirmed attractants), already had been identified for most of these species (summarized in Hanks and Millar 2016). For example, the (R)-3-hydroxyhexan-2-one enantiomer of 3-ketol is a dominant or sole pheromone component of the cerambycine species (ordered as in Table 2): Phymatodes aereus (Newman), Neoclytus m. mucronatus (F.), and Xylotrechus colonus (F.). Due to earlier taxonomic confusion between Phymatodes amoenus (Say) and Phymatodes lengi Joutel, the pheromone of the former species has been erroneously identified as being composed of (R)-3-hydroxyhexan-2-one and (R)-2-methylbutan-1-ol (Mitchell et al. 2015, Hanks and Millar 2016). The pheromone of P. amoenus now is known to comprise only (R)-2-methylbutan-1-ol, which accounts for attraction of that species to racemic 2-methylbutan-1-ol in the present study ('2-Me-ol' in Table 2). The pheromone of the exotic congener Phymatodes testaceus (L.) also is composed solely of (R)-2-methylbutan-1-ol (Hanks and Millar 2016), to which it was attracted in the present study. Attraction of Phymatodes dimidiatus (Kirby) by 2-methylbutan-1-ol represents new information about its likely pheromone. Last, the cerambycine Neoclytus acuminatus acuminatus (F.) was attracted to the syn-diol treatment as expected, because it contained the pheromone (2S,3S)-2,3-hexanediol.

Some species of lamiines were attracted to the known lamiine pheromone compounds, as would be expected from previous research (Hanks and Millar 2016), with fuscumol acetate ('Fusc. acet.' in Table 2) attracting Graphisurus fasciatus (Degeer), Lepturges angulatus (LeConte), and Aegomorphus modestus (Gyllenhal), fuscumol ('Fusc.') attracting Sternidius alpha (Say), and monochamol ('Monoch.') attracting Monochamus carolinensis (Olivier) and its congener Monochamus scutellatus scutellatus (Say). A third congener, Monochamus notatus (Drury) was most strongly attracted to the pheromone blend in New York, with intermediate attraction to monochamol. However, an earlier field study, conducted at the same study sites in New York, showed that adults of M. notatus were significantly attracted to monochamol alone (Fierke et al. 2012). The lamiine Styloleptus b. biustus (LeConte) was attracted by the blend of fuscumol and fuscumol acetate in Florida, and both of the Tetropium species (subfamily Spondylidinae) by fuscumol alone, consistent with results reported in a complementary study (Millar et al. 2017). Attraction of Oplosia nubila (LeConte) to fuscumol acetate represents new information about its possible pheromone component.

Comparing the responses of the various species to the blend of six pheromones versus the individual components illustrated both synergistic and antagonistic effects among the blend components. The blend was significantly less attractive, or not at all attractive to several species, compared to the individual components (Table 2), suggesting that one or more components of the blend had inhibited responses. Notable examples are the strong antagonistic effect of the 3-ketol component of the blend on attraction of P. amoenus to its pheromone (R)-2-methylbutan-1-ol, and N. a. acuminatus to its pheromone (2S,3S)-2,3-hexanediol (Hanks et al. 2012, Miller et al. 2017; L.M.H., unpub. data). Other species showed similar evidence of antagonism, including the cerambycines P. aereus and N. m. mucronatus, and the lamiines S. alpha, S. b. biustus, A. modestus, O. nubila, and M. s. scutellatus. For the spondylidine species in the genus Tetropium, the blend attracted significant numbers of beetles, although fewer beetles than were attracted by fuscumol alone, suggesting some degree of antagonism by other blend components.

In contrast, there was evidence of synergism between components in the blend for only two species. Thus, the greater attraction of *X. colonus* to the blend is no doubt due to the pairing of *syn*-diol, a minor component of its pheromone, with 3-ketol, the dominant component (Lacey et al. 2009). Similarly, adult males of *Anelaphus inermis* (Newman) produce the (2S,3R)-enantiomer of the *anti*diol along with a small amount of *syn*-diol, which accounts for its stronger attraction to the complete blend of chemicals versus the individual compounds.

The remaining species were similarly attracted to their individual pheromone (or attractant) components and to the complete blend, suggesting that the components of the blend that were not part of their pheromones neither synergized nor antagonized attraction. These species included the cerambycines *P. dimidiatus* and *P. testaceus*, and the lamiines *G. fasciatus*, *L. angulatus*, and the three *Monochamus* species.

Taken together, traps baited with the blend of six pheromone components attracted significant numbers of 10 species or about half of the cerambycid species in Table 2, but antagonism significantly reduced or even eliminated attraction of the remaining 9 species. Nevertheless, the blend was a better general attractant than any of the individual components. For example, 3-ketol attracted only three species of cerambycines and no lamiines, and fuscumol acetate attracted only four species of lamiines and no cerambycines.

In experiment 2, there were no cases in which the plant volatiles ethanol and α-pinene antagonized attraction of a species to the pheromone blend. Several species were attracted to traps baited only with the plant volatiles (Table 3). Among the few species that were attracted by plant volatiles alone, the pheromone blend strongly antagonized attraction of the cerambycine Anelaphus villosus (F.). Although the two spondylidine Tetropium species also were attracted by plant volatiles alone, attraction possibly was enhanced by fuscumol in the pheromone blend (Table 3) based on previous reports of synergism between their pheromones and host plant volatiles (Silk et al. 2007; Sweeney et al. 2010, 2014). Similarly, the spondylidine Asemum striatum (L.) was significantly attracted by the combination of the pheromone blend with plant volatiles, but not to either alone, as previously reported (Collignon et al. 2016). This species also was attracted by the blend of fuscumol and fuscumol acetate in the companion article (Millar et al. 2017), but the present findings suggest that attraction to those chemicals in the blend was antagonized by other components. All six of the Monochamus species and subspecies were attracted only by the combination of pheromone blend and plant volatiles, the key blend component being monochamol (Millar and Hanks 2017). Plant volatiles are known to be important synergists of monochamol for at least some species of Monochamus (e.g., Teale et al. 2011, Allison et al. 2012, Hanks et al. 2012). Lastly, nothing is known of the pheromone chemistry of the cerambycine Euderces picipes (F.) but attraction only to the combination of the pheromone blend and plant volatiles suggests that the blend contains one or more components of its pheromone.

For the remaining species in experiment 2, plant volatiles apparently did not influence their responses to particular components of the pheromone blend (Table 3), including the cerambycines *X. colonus* (probably responding to 3-ketol and diols in the blend), the lamiines *G. fasciatus* (responding to fuscumol acetate), and the lamiine *S. b. biustus* (responding to fuscumol + fuscumol acetate; Table 3). These findings were consistent with an earlier study (Hanks and Millar 2013). Nevertheless, Miller et al. (2015a,b) found that ethanol alone enhanced attraction of *X. colonus* to 3-ketol.

In summary, the experiments described here illustrate two important points. First, it is clear that blends of cerambycid pheromones can be deployed to catch multiple species of cerambycids

| $dman's Q (df)' = Blend' = 3.Ketol = $.yn-Diol$ $dman's Q (df)' = Blend' = 3.Ketol = $.yn-Diol$ $0.17 \pm 0.177 \pm 0.177 = 0.177 = 0.177 = 0.177 = 0.177 = 0.177 = 0.177 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.161 = 0.05 = 0.175 = 0.35 \pm 0.150 = 0.03 \pm 0.050 = 0.03 \pm 0.050 = 0.08 = 0.050 = 0.08 = 0.050 = 0.08 \pm 0.050 = 0.08 = 0.050 = 0.08 \pm 0.050 = 0.08 = 0.050 = 0.08 = 0.050 = 0.08 = 0.050 = 0.08$ | | 2-Me-ol | $\mathrm{Fusc.}^{e}$ | ; | | |
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| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | | 0b | 0b | 0b | 0b | 0 |
| $10.8 \pm 8.4ab 0.25 \pm 0.16b 0b \\ 2.3 \pm 1.1a 0b 0b \\ 1.2 \pm 0.6ab 0.6 \pm 0.4ab 0b \\ 1.2 \pm 0.6ab 0.6 \pm 0.4ab 0b \\ 0.39 \pm 0.17b 0.64 \pm 0.4b 9.2 \pm 1.4a \\ 0.5 \pm 0.17b 0.64 \pm 0.4b 9.2 \pm 1.4a \\ 0.5 \pm 1.5b 3.4 \pm 3b 2.95 \pm 7.8a \\ 5.3 \pm 1.5b 3.4 \pm 3b 2.95 \pm 7.8a \\ 0.75 \pm 0.5b 7.0 \pm 2.9a 0b \\ 2.1 \pm 0.4b 12.5 \pm 4.1a 5.5 \pm 1.8b \\ 5.0 \pm 0.9a 3.0 \pm 0.7b 1.1 \pm 0.38bc \\ 7.6 \pm 3.1a 4.7 \pm 2.3ab 1.5 \pm 1.5bc \\ 3.8 \pm 0.9a 3.0 \pm 0.7b 0.08 \pm 0.06c \\ 1.2 \pm 0.3a 0.22 \pm 0.15b 0.08 \pm 0.06c \\ 1.2 \pm 0.3a 0.22 \pm 0.15b 0.08 \pm 0.06c \\ 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b 0 \\ 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b \\ 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b \\ 0 \end{bmatrix}$ | | 0b | 0b | 0b | 0b | $0.33 \pm 0.3b$ |
| $10.8 \pm 8.4ab 0.25 \pm 0.16b 0b$ $2.3 \pm 1.1a 0b 0.6 \pm 0.4ab 0b$ $1.2 \pm 0.6ab 0.6 \pm 0.4ab 0b$ $0.36 \pm 0.17b 0.64 \pm 0.4b 9.2 \pm 1.4a$ $0b 0b 3.4 \pm 0.9a$ $5.3 \pm 1.5b 3.4 \pm 3b 29.5 \pm 7.8a$ $0.75 \pm 0.5b 7.0 \pm 2.9a 0b$ $2.1 \pm 0.4b 12.5 \pm 4.1a 5.5 \pm 1.8b$ $5.0 \pm 0.9a 3.0 \pm 0.7b 1.1 \pm 0.38bc$ $7.6 \pm 3.1a 4.7 \pm 2.3ab 1.5 \pm 1.5bc$ $3.8 \pm 0.9a 2.2 \pm 0.6b 0.08 \pm 0.06c$ $1.2 \pm 0.3a 0.22 \pm 0.15b 0.08 \pm 0.06c$ $1.2 \pm 0.3a 0.22 \pm 0.15b 0b$ $1.1 \pm 0.3ab 0.33 \pm 0.2b 0b$ $1.3 \pm 0.5a 0.38 \pm 0.2b 0b$ $1.3 \pm 0.5a 0.5b 0b$ $1.3 \pm 0.5a 0.5b 0b$ $1.3 \pm 0.5a 0.5b 0b$ $1.4 \pm 2.6a 0.38 \pm 0.2b 0b$ $1.5 \pm 0.5b 0b$ 1 | | | | | | |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | | $31.5 \pm 17a$ | $3.1 \pm 2.9b$ | $0.5 \pm 0.5b$ | $1.6 \pm 1.5b$ | 0 |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 2 | | 2 | 5 | 5 | 5 |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | 0p | $1.5 \pm 0.6a$ | 0p | 0b | 0b | 0b |
| $1.2 \pm 0.6 \text{ ab} \qquad 0.6 \pm 0.4 \text{ ab} \qquad 0.6$ $0.39 \pm 0.17 \text{ b} \qquad 0.72 \pm 0.39 \qquad 2.7 \pm 0.6 \text{ a} \\ 0.36 \pm 0.17 \text{ b} \qquad 0.64 \pm 0.4 \text{ b} \qquad 9.2 \pm 1.4 \text{ a} \\ 0.35 \pm 0.17 \text{ b} \qquad 0.64 \pm 0.4 \text{ b} \qquad 9.2 \pm 1.4 \text{ a} \\ 0.75 \pm 0.5 \text{ b} \qquad 3.4 \pm 3 \text{ b} \qquad 29.5 \pm 7.8 \text{ a} \\ 2.1 \pm 0.4 \text{ b} \qquad 12.5 \pm 4.1 \text{ a} \qquad 5.5 \pm 1.8 \text{ b} \\ 2.1 \pm 0.4 \text{ b} \qquad 12.5 \pm 4.1 \text{ a} \qquad 5.5 \pm 1.8 \text{ b} \\ 7.6 \pm 3.1 \text{ a} \qquad 4.7 \pm 2.3 \text{ ab} \qquad 1.1 \pm 0.38 \text{ b} \\ 7.6 \pm 3.1 \text{ a} \qquad 2.2 \pm 0.6 \text{ b} \qquad 0.08 \pm 0.06 \text{ c} \\ 1.2 \pm 0.3 \text{ a} \qquad 0 \text{ b} \qquad 0.2 \pm 0.1 \text{ b} \\ 1.2 \pm 0.3 \text{ a} \qquad 0 \text{ b} \qquad 0.2 \pm 0.1 \text{ b} \\ 1.1 \pm 0.3 \text{ b} \\ 1.1 \pm 0.3 \text{ b} \qquad 0.33 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ a} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ a} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ a} \qquad 0 \text{ b} \qquad 0.2 \pm 0.15 \text{ b} \\ 1.4 \pm 2.6 \text{ a} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ a} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ a} \qquad 0 \text{ b} \qquad 0 \text{ b} \\ 1.3 \pm 0.5 \text{ c} \qquad 0 \text{ b} \\ 0 \text{ b} \\ 4.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.38 \pm 0.2 \text{ b} \qquad 0 \text{ b} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.4 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.4 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.4 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0.4 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.4 \pm 2.6 \text{ c} \qquad 0 \text{ c} \\ 1.$ | | 0 0 0 | į | į | Ţ | 1 |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | | $2.0 \pm 0.8a$ | 0b | 0b | 0b | 0b |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | | | | | | |
| *** $0.39 \pm 0.1b$ $0.72 \pm 0.3b$ $2.7 \pm 0.6a$ ** $0.36 \pm 0.17b$ $0.64 \pm 0.4b$ $9.2 \pm 1.4a$ * $5.3 \pm 1.5b$ $3.4 \pm 3b$ $2.9.5 \pm 7.8a$ * $5.3 \pm 1.5b$ $3.4 \pm 3b$ $2.9.5 \pm 7.8a$ * $0.75 \pm 0.5b$ $7.0 \pm 2.9a$ $0b$ * $2.11 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ * $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.1 \pm 0.38bc$ *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.1 \pm 0.38bc$ *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ *** $1.2 \pm 0.3a$ $0b$ $0.08 \pm 0.06c$ *** $1.2 \pm 0.3a$ $0b$ $0.03 \pm 0.06c$ *** $1.2 \pm 0.3a$ $0.33 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.33 \pm 0.2b$ $0b$ | | | | | | |
| ** $0.36 \pm 0.17b$ $0.64 \pm 0.4b$ $9.2 \pm 1.4a$ $0b$ $0b$ $3.4 \pm 3b$ $2.9.5 \pm 7.8a$ <i>attus</i> $0.75 \pm 0.5b$ $3.4 \pm 3b$ $2.9.5 \pm 7.8a$ $1.1 \pm 0.75 \pm 0.5b$ $7.0 \pm 2.9a$ $0b$ $3.4 \pm 0.0b$ * $2.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ $2.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ $1.1 \pm 0.38bc$ $0.08 \pm 0.06c$ *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.1 \pm 0.38bc$ *** $3.8 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ *** $1.2 \pm 0.3a$ $0b$ $0.08 \pm 0.06c$ *** $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.33 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | $0.03 \pm 0.03b$ | $0.08 \pm 0.05b$ | $0.06 \pm 0.05b$ | $0.056 \pm 0.04b$ | $0.03 \pm 0.03 b$ |
| ** 0b $0b$ $3.4 \pm 3b$ $3.4 \pm 0.9a$ ** $5.3 \pm 1.5b$ $3.4 \pm 3b$ $29.5 \pm 7.8a$ natus $0.75 \pm 0.5b$ $7.0 \pm 2.9a$ $0b$ ** $2.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ ** $5.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ ** $5.0 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ ** $5.0 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ ** $5.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.1 \pm 0.38bc$ ** $3.8 \pm 0.9a$ $3.0 \pm 0.7b$ $0.08 \pm 0.06c$ *** $3.8 \pm 0.9a$ $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ *** $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ $0b$ ** $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.38 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.38 \pm 0.2b$ $0b$ | | $0.07 \pm 0.07b$ | 0b | $0.14 \pm 0.1b$ | $0.07 \pm 0.07b$ | 0b |
| * 5.3 \pm 1.5b 3.4 \pm 3b 29.5 \pm 7.8a <i>natus</i> 0.75 \pm 0.5b 7.0 \pm 2.9a 0b * 2.1 \pm 0.4b 12.5 \pm 4.1a 5.5 \pm 1.8b 2.1 \pm 0.4b 12.5 \pm 4.1a 5.5 \pm 1.8b * 5.0 \pm 0.9a 3.0 \pm 0.7b 1.1 \pm 0.38bc 7.6 \pm 3.1a 4.7 \pm 2.3ab 1.5 \pm 0.08 \pm 0.06c *** 3.8 \pm 0.9a 2.2 \pm 0.6b 0.08 \pm 0.06c *** 1.2 \pm 0.3a 0.22 \pm 0.1b * 1.2 \pm 0.3a 0.22 \pm 0.1b 0b * 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b * 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b * 1.3 \pm 0.5a 0.38 \pm 0.2b 0b * 4.4 \pm 2.6a 0.38 \pm 0.2b 0b | | 0b | 0b | 0b | 0b | 0b |
| aatus $0.75 \pm 0.5b$ $7.0 \pm 2.9a$ $0b$ * $2.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ ** $5.0 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ ** $5.0 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ ** $5.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ ** $3.8 \pm 0.9a$ $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ ** $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.38 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.38 \pm 0.2b$ $0b$ | 29.5 ± 7.8a | 0b | 0b | 0b | 0b | $0.13 \pm 0.1b$ |
| * $0.75 \pm 0.5b$ $7.0 \pm 2.9a$ $0b$ 2.1 \pm 0.4b $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ 2.1 \pm 0.4b $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ 5.0 \pm 0.9a $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ 7.6 \pm 3.1a $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ 3.8 \pm 0.9a $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ 1.2 \pm 0.3a $0.2 \pm 0.1b$ $0.2 \pm 0.1b$ 1.1 \pm 0.3ab $0.23 \pm 0.2b$ $0b$ 1.1 \pm 0.3ab $0.33 \pm 0.2b$ $0b$ 1.1 \pm 0.3ab $0.33 \pm 0.2b$ $0b$ 4.4 \pm 2.6a $0.38 \pm 0.2b$ $0b$ | | | | | | |
| * $2.1 \pm 0.4b$ $12.5 \pm 4.1a$ $5.5 \pm 1.8b$ 5.0 \pm 0.9a $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ 3.8 \pm 0.9a $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ *** $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.1 \pm 0.3ab$ $0.2 \pm 0.1b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | 0b | 0b | $0.5 \pm 0.5b$ | 0b | 0 |
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| ** $5.0 \pm 0.9a$ $3.0 \pm 0.7b$ $1.1 \pm 0.38bc$ *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ **** $3.8 \pm 0.9a$ $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ **** $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.1 \pm 0.3ab$ $0.22 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.2b$ $0b$ * $1.3 \pm 0.5a$ $0.33 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | | | | | |
| *** $7.6 \pm 3.1a$ $4.7 \pm 2.3ab$ $1.5 \pm 1.5bc$ **** $3.8 \pm 0.9a$ $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ 1.2 $\pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.2b$ $0b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | $0.19 \pm 0.06c$ | $0.44 \pm 0.26c$ | $0.37 \pm 0.2c$ | $1.2 \pm 0.6 bc$ | $0.09 \pm 0.09c$ |
| *** $3.8 \pm 0.9a$ $2.2 \pm 0.6b$ $0.08 \pm 0.06c$ * $1.2 \pm 0.3a$ $0b$ $0.2 \pm 0.1b$ * $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.23 \pm 0.2b$ $0b$ * $1.3 \pm 0.5a$ $0b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | $0.33 \pm 0.2c$ | 0c | $0.13 \pm 0.1c$ | 0c | $0.33 \pm 0.2c$ |
| * $1.2 \pm 0.3a$ 0b $0.2 \pm 0.1b$ 8 $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ 0b 1.1 \pm 0.3ab $0.33 \pm 0.2b$ 0b 1.3 \pm 0.5a 0b 4.4 \pm 2.6a 0.38 \pm 0.2b 0b | | $0.08 \pm 0.06c$ | $0.04 \pm 0.04c$ | $0.04 \pm 0.04c$ | $0.04 \pm 0.04c$ | $0.12 \pm 0.09c$ |
| * $1.2 \pm 0.3a$ 0b $0.2 \pm 0.1b$ * $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ 0b 1.1 \pm 0.3ab $0.33 \pm 0.2b$ 0b 1.1 \pm 0.3ab 0.33 \pm 0.2b 0b * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ 0b | | | | | | |
| * $1.2 \pm 0.3a$ 0b $0.2 \pm 0.1b$ s $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ 0b 1.1 \pm 0.3ab 0.22 \pm 0.15b 0b 1.1 \pm 0.3ab 0.23 \pm 0.2b 0b * $1.3 \pm 0.5a$ 0b 0b * $4.4 \pm 2.6a$ 0.38 \pm 0.2b 0b | | | | | | |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | $0.2 \pm 0.1b$ | $0.1 \pm 0.1b$ | 0.3 | $0.3 \pm 0.15b$ | 0b | 0b |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | | | | | | |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | | | | | | |
| * $1.0 \pm 0.3a$ $0.22 \pm 0.15b$ $0b$ * $1.1 \pm 0.3ab$ $0.33 \pm 0.2b$ $0b$ * $1.3 \pm 0.5a$ $0b$ $0b$ * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ $0b$ | | | | | | |
| * $1.1 \pm 0.3 ab$ $0.33 \pm 0.2 b$ $1.3 \pm 0.5 a$ $0b$ $4.4 \pm 2.6 a$ $0.38 \pm 0.2 b$ | | $0.22 \pm 0.15 ab$ | $0.11 \pm 0.1b$ | $1.0 \pm 0.4a$ | $0.44 \pm 0.2b$ | $0.11 \pm 0.1b$ |
| * $1.3 \pm 0.5a$ 0b $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ | | $0.09 \pm 0.09b$ | 0b | $2.1 \pm 0.85a$ | 0b | $0.27 \pm 0.14b$ |
| * $1.3 \pm 0.5a$ 0b * $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ | | | | | | |
| $()^{***}$ $4.4 \pm 2.6a$ $0.38 \pm 0.2b$ | 0b | 0b | 0b | $0.50 \pm 0.3b$ | 0b | 0b |
| | | 0b | 0b | $6.4 \pm 2.5a$ | 0b | $0.13 \pm 0.1b$ |
| 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 | | | | | | |
| $1.7 \pm 0.9b$ $0.1 \pm 0.1b$ | b $0.2 \pm 0.1b$ | 0 | $7.4 \pm 2.9a$ | $0.3 \pm 0.2b$ | 0p | $0.1 \pm 0.1b$ |
| S | | | | | | |
| FL: $24.2 (6,70)^{**}$ 0.2 ± 0.1b 0.4 ± 0.2b 0b | | $0.1 \pm 0.1b$ | 5. | 2.1 ± 0.6a | 0b | $0.3 \pm 0.2b$ |

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| Taxonomy/ State/ Friedman's Q (df) ^a Blend ^b Acanthoderini Aegomorphus modestus DN: 30 2 77 641*** | | | | | | | | |
|--|------------------------|-------------------------------|---------------------------|--------------------------|-------------------------|-----------------------|------------------------------------|-------------------|
| | nd ^b | 3-Ketol | syn-Diol | 2-Me-ol | $Fusc.^{e}$ | Fusc. acet. | Monoch. | Control |
| | | | | | | | | |
| | | | | | | | | |
| | $0.38 \pm 0.38b$ | $0.13 \pm 0.1b$ | 0b | 0b | 0b | $2.0 \pm 0.53a$ | 0b | $0.25 \pm 0.2b$ |
| MI: 44.2 $(7,64)^{***}$ 0.9 ± 0 | $0.9 \pm 0.55b$ | 0b | 0 | 0 | $0.13 \pm 0.13b$ | $1.8 \pm 0.6a$ | 0b | 0b |
| NY: $20.0 (7,59)^{**}$ $0.50 \pm 0.3b$ | 0.3b | 0b | 0b | $0.14 \pm 0.1b$ | 0b | $2.5 \pm 1.5a$ | $0.29 \pm 0.2b$ | 0b |
| Oplosia nubila | | | | | | | | |
| IN: 32.2 (7,48)*** 0b | q | 0b | 0b | 0b | $0.17 \pm 0.17b$ | $1.2 \pm 0.4a$ | 0b | 0 |
| Monochamini | | | | | | | | |
| Monochamus carolinensis | | | | | | | | |
| TX: $22.5 (7,32)^{**}$ 4.0 ± 0.9a | : 0.9a | 0b | 0b | $0.25 \pm 0.2b$ | $0.25 \pm 0.2b$ | $3.0 \pm 2ab$ | $5.0 \pm 1.7a$ | 0 |
| Monochamus notatus | | | | | | | | |
| NY: $35.8 (7,80)^{***}$ 0.9 ± 0.3a | : 0.3a | 0b | 0b | 0b | 0b | 0b | $0.5 \pm 0.2ab$ | 0b |
| Monochamus s. scutellatus | | | | | | | | |
| MI: $20.9 (7,32)^{**}$ 1.0 ± 0.6b | : 0.6b | 0b | 0b | $^{0}\mathrm{p}$ | 0b | $0.5 \pm 0.3 b$ | $5.3 \pm 3.6a$ | 0 |
| NY: $41.1 (7, 176)^{***}$ 1.2 ± 0.3a | : 0.3a | $0.09 \pm 0.6b$ | $0.045 \pm 0.4b$ | $0.046 \pm 0.04b$ | $0.045 \pm 0.4b$ | $0.045 \pm 0.4b$ | $1.7 \pm 0.3a$ | $0.09 \pm 0.06b$ |
| Spondylidinae | | | | | | | | |
| Asemini | | | | | | | | |
| Tetropium cinnamopterum | | | | | | | | |
| NY: $106.1 (7,264)^{***}$ 1.9 ± 0 | $1.9 \pm 0.46b$ | 0c | $0.09 \pm 0.09c$ | $0.06 \pm 0.04c$ | $3.4 \pm 0.8a$ | $0.09 \pm 0.09c$ | $0.09 \pm 0.09c$ | $0.12 \pm 0.1c$ |
| Tetropium schwarzianum | | | | | | | | |
| NY: $70.6 (7, 184)^{***}$ 0.70 ± 0.2b | : 0.2b | 0c | $0.04 \pm 0.04c$ | $0.04 \pm 0.04c$ | $1.2 \pm 0.3a$ | $0.09 \pm 0.06c$ | 0c | $0.09 \pm 0.06c$ |
| | | | | | | | | |
| Means in bold were significantly different from controls. Abbreviations for chemicals: Blend of all the following chemicals; 3-Ketol = racemic 3-hydroxyhexan-2-one; syn-Diol = syn-2, 3-hexanediol; 2-Me-ol = racemic | ols. Abbrevia | tions for chemicals: Ble | nd = blend of all the fol | llowing chemicals; 3-Ke | stol = racemic 3-hydrox | yhexan-2-one; syn-Dio | <pre>l = syn-2,3-hexanediol;</pre> | 2-Me-ol = racemic |
| 2-methybutan-1-ol; Fusc. = racemic fuscumol; Fusc. Acet. = racemic fuscumol acetate; Monoch. = monochamol; Control = solvent control. | et. = racemic | fuscumol acetate; Mor | och. = monochamol; C | control = solvent contro | ol. | | | |
| ^a Asterisks indicate significance level of Friedman's Q: *P < 0.01; **P < 0.001; ***P < 0.0001. | : * <i>P</i> < 0.01; * | $^{*}P < 0.001; ^{***}P < 0.$ | 0001. | | | | | |
| ^b Means within species with different letters are significantly different (REGWQ test, $P < 0.05$). | ficantly differe | ent (REGWQ test, $P <$ | 0.05). | | | | | |

"Merged cells for fuscumol and fuscumol acetate for the Florida site indicates that only the blend had been tested, not the individual compounds.

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| Taxonomy/State/Friedman's $Q (df)^1$ | Pheromone blend ² | Blend + Plant volatiles | Plant volatiles | Solvent control |
|--------------------------------------|------------------------------|-------------------------|-----------------|-----------------|
| Cerambycinae | | | | |
| Clytini | | | | |
| Xylotrechus colonus | | | | |
| NY: 25.0 (3,88)*** | $1.4 \pm 0.3a$ | $1.7 \pm 0.4a$ | $0.32 \pm 0.2b$ | $0.36 \pm 0.3b$ |
| Elaphidiini | | | | |
| Anelaphus villosus | | | | |
| MI: 15.4 (3,24)** | 0b | $0.5 \pm 0.3b$ | $4.0 \pm 1.1a$ | $0.33 \pm 0.2b$ |
| Tillomorphini | | | | |
| Euderces picipes | | | | |
| MI: 16.3 (3,28)** | $0.7 \pm 0.5 b$ | $4.0 \pm 1.4a$ | $0.43 \pm 0.4b$ | 0b |
| Lamiinae | | | | |
| Acanthocinini | | | | |
| Graphisuris fasciatus | | | | |
| NY: 11.9** | $2.4 \pm 0.6a$ | $1.8 \pm 0.6a$ | 0b | $0.4 \pm 0.4b$ |
| Styloleptus b. biustus | | | | |
| FL: 16.2 (3,40)** | $4.5 \pm 1.0a$ | 1.8 ± 1.1ab | $0.2 \pm 0.1b$ | $0.2 \pm 0.2b$ |
| Monochamini | | | | |
| Monochamus carolinensis | | | | |
| MI: 11.6 (3,20)** | 1.4 ± 0.9ab | $7.8 \pm 3.9a$ | 4.0 ± 1.7ab | $0.2 \pm 0.2b$ |
| Monochamus clamator | | | | |
| CA: 22.4 (3,24)*** | 0b | $2.5 \pm 0.4a$ | 0b | 0b |
| Monochamus notatus | | | | |
| NY: 44.0 (3,85)*** | $0.62 \pm 0.2b$ | $4.4 \pm 0.8a$ | $0.67 \pm 0.3b$ | $0.5 \pm 0.48b$ |
| Monochamus obtusus | | | | |
| CA: 20.7 (3,44)*** | $3.8 \pm 2.0b$ | $27.5 \pm 9.2a$ | $1.3 \pm 0.7b$ | $0.1 \pm 0.1b$ |
| Monochamus s. oregonensis | | | | |
| CA - 15.7 (3,20)** | $0.2 \pm 0.2b$ | $2.0 \pm 0.6a$ | 0b | 0b |
| Monochamus s. scutellatus | | | | |
| NY: 39.2 (3,92)*** | $1.1 \pm 0.4b$ | $10.5 \pm 2.4a$ | $1.4 \pm 0.4b$ | $0.43 \pm 0.2b$ |
| Spondylidinae | | | | |
| Asemini | | | | |
| Asemum striatum | | | | |
| NY: 29.0 (3,72)*** | $0.5 \pm 0.1b$ | $2.5 \pm 0.6a$ | $0.8 \pm 0.4b$ | 0b |
| Tetropium cinnamopterum | | | | |
| NY: 29.2 (3,120)*** | 2.6 ± 0.5bc | $8.1 \pm 2.1a$ | $4.6 \pm 1.2b$ | $1.2 \pm 0.8c$ |
| Tetropium schwarzianum | | | | |
| NY: 32.0 (3,112)*** | $1.3 \pm 0.3 bc$ | $3.6 \pm 0.6a$ | $2.1 \pm 0.5b$ | $0.4 \pm 0.3c$ |

 Table 3. Mean (±1 SE) number of cerambycid beetles captured per treatment and replicate during Experiment 2, and results of Friedman's Q analysis, for the cases in which overall treatment effects were statistically significant

Treatments included the pheromone blend alone (components listed in footnote of Table 2), the pheromone blend + plant volatiles (ethanol and α -pinene), plant volatiles alone, and solvent control.

¹Asterisks indicate significance level of Friedman's Q: ** P < 0.001; *** P < 0.0001.

²Means within species with different letters are significantly different (REGWQ test, P < 0.05).

simultaneously, but it is equally clear that the choice of which compounds to combine can be critically important. Our data suggest that attraction of a given species to a synthesized reconstruction of its pheromone is most likely to be antagonized by compounds produced by close relatives. Examples include cerambycine species which are attracted by ketol components of their pheromones but antagonized by diols produced by other species of cerambycines, and vice versa (e.g., Millar et al. 2017). In contrast, it is unlikely that attraction would be antagonized by pheromone components of more distantly related species. For example, we know of no cases in which the lamiine and spondylidine pheromones fuscumol or fuscumol acetate have been shown to inhibit attraction of cerambycines to ketols or diols, and vice versa. Choices of possible future blends for testing should be guided by these two principles. It is likely that the optimal strategy, in terms of attracting the greatest diversity of cerambycid species, would be to use multiple traps, each baited with different blends which have been crafted to minimize antagonism among components. In comparison to using traps baited with single

compounds, or blends optimized for a single species, such composite blends would still reduce the number of traps necessary to cast a broad net and so minimize costs of materials and labor. Second, there may be considerable advantages to deploying

simple blends of host plant volatiles such as ethanol and α -pinene with pheromones, in terms of increasing both the number of species and the number of individuals attracted. In the studies reported here, there was no indication that ethanol and α -pinene antagonized attraction of any species to pheromone lures. In another large study, Miller et al. (2017) similarly found that ethanol often enhanced attraction of cerambycids to pheromone lures, with no evidence of inhibition. However, Collignon et al. (2016) reported that attraction of some cerambycid species that infest hardwoods to their pheromones was interrupted by high release rates of volatiles that are typical of conifers. Thus, even with host plant compounds, the anticipated target species and their typical host plants must be considered when choosing which pheromones and host plant volatiles to formulate into optimal attractants for multiple species simultaneously.

Supplementary Data

Supplementary data are available at *Journal of Economic Entomology* online.

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